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Peroxynitrite scavenging activity of lithospermate B from *Salvia miltiorrhiza*

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Abstract

Peroxynitrite (ONOO⁻) is produced by the reaction of superoxide (O_2^-) with nitric oxide. ONOO⁻ damages proteins through nitration or oxidation. For protection from ONOO⁻ induced protein modifications, ONOO⁻ scavengers should be supplemented. Evidence was obtained that lithospermate B extracted from *Salvia miltiorrhiza* showed the strongest scavenging activity among its constituents. Its ONOO⁻ scavenging activity is via an electron donation mechanism. A dihydroxyl group and a double bond seem to be essential structure requirements. The data from the experiments further confirmed a protective effect of lithospermate B on bovine serum albumin and low-density lipoprotein against ONOO⁻. This study demonstrated that lithospermate B with hydroxyl groups and double bonds exerts an anti-nitration effect by scavenging ONOO⁻.

Introduction

Oxidative stress is known to occur widespread throughout the body (Gutteridge 1999). Peroxynitrite (ONOO⁻), a product of nitrogen-derived oxidative stress, is formed by the reaction of two ubiquitous free radical species, superoxide (O_2^-) and nitric oxide (NO) (Pryor and Squadrito 1995). ONOO⁻ is a powerful oxidant that can cause damage to proteins through a subtle mechanism called nitration (Arteel et al 1999). Tyrosine has been proven to be extremely sensitive to ONOO⁻. In fact, nitrotyrosine has served as a fingerprint for ONOO⁻ mediated damage of cellular proteins in a variety of pathological conditions, including ischaemia/reperfusion injury, Alzheimer's disease and atherosclerosis (Smith et al 1997; He et al 1999; Luoma & Yla-Herttuala 1999).

Low molecular mass compounds such as glutathione, methionine, cysteine, melatonin and serotonin have been found to neutralize cytotoxic ONOO⁻ (Gilad et al 1997; Sandoval et al 1997; Soung et al 1998). In addition, NO and O_2^- levels are shown to be controlled by oxyhaemoglobin and superoxide dismutase, respectively (Gilad et al 1997). However, in order to overcome excessive ONOO⁻ produced during pathological conditions, scavengers are needed. Scavengers for ONOO⁻ include the antioxidant ingredients found in natural products such as flavonoids and hydroxycinnamates (Haenen et al 1997; Pannala et al 1998). We have previously reported on the ONOO⁻ scavenging effects of catechin polyphenols extracted from green tea and the phenolic compounds from *Eriobotrya japonica* (Chung et al 1998; Soung et al 1999).

Extracts from *Salvia miltiorrhiza* consisting of phenolic compounds have been widely used in China as herbal medicine for the clinical treatment of vascular system abnormalities (Tang & Eisenbrand 1992). The major active constituents of *S. miltiorrhiza* are identified as caffeic acid, trans-rosmarinic acid, lithospermate A, lithospermate B, magnesium lithospermate B, and ammonium-potassium lithospermate B (Tanaka et al 1989). However, to date, the ONOO⁻ scavenging activity of *S. miltiorrhiza* and its major component, lithospermate B, has not been reported.

In the present study, we investigated the $ONOO^-$ scavenging effect of *S. miltior-rhiza* and its fractions using six components extracted from the subfraction A-2 in *S. miltiorrhiza*. We found lithospermate B to be the most potent $ONOO^-$ scavenger. In addition, we were able to estimate the scavenging activity of lithospermate B against $ONOO^-$ and the ability of lithospermate B to protect tyrosine, a cellular component,

against ONOO⁻ by an electron donation mechanism. We also examined the possible protective effect of lithospermate B against ONOO⁻ induced nitration of bovine serum albumin (BSA) and low-density lipoprotein (LDL).

Materials and Methods

Materials

S. miltiorrhiza and its components (caffeic acid, transrosmarinic acid, lithospermate A, lithospermate B, magnesium lithospermate B and ammonium-potassium lithospermate B) were isolated and provided by Dr Itsuo Nishioka, Kyushu University, Japan. ONOO⁻ and dihydrorhodamine 123 (DHR 123) were obtained from Cayman Chemical Company (Ann Arbor, MI, USA) and Molecular Probes (Eugene, OR, USA), respectively. BSA and LDL were purchased from Sigma-Aldrich Chemical Company (St Louis, MO, USA). Anti-nitrotyrosine antibody and horseradish peroxidase-conjugated antimouse secondary antibody from sheep were obtained from Upstate Biotechnology (Lake Placid, NY, USA) and Amersham (Buckinghamshire, UK), respectively.

Isolation of caffeic acid analogues from *S. miltiorrhiza* Radix

Commercially available S. miltiorrhiza Radix (1.0 kg) produced in China was extracted twice with water (1.5 L) at 80°C (fraction A) and then extracted twice with acetone (1.52 L) at room temperature (fraction B). After removal of the insoluble fraction by filtration, the filtrate was concentrated under reduced pressure (ca 40 °C) and subjected to MCI-gel CHP 20 P (7.5 cm i.d. \times 35 cm) column chromatography. After washing the column with water, the samples were chromatographed over Sephadex LH-20 (5.0 cm i.d. \times 42 cm) with water/EtOH to afford subfractions A-1, A-2 and A-3. Magnesium lithospermate B (7.56 g) was isolated from subfraction A-2. Further elution of the column with water containing increasing proportions of ethanol yielded rosmarinic acid (0.15 g), caffeic acid (0.03 g) and lithospermic acid B (1.86 g). Fraction 1 was dissolved in 0.5 M hydrochloric acid (5 mL), applied directly to a Sephadex LH-20 column $(3.5 \text{ cm i.d.} \times 30 \text{ cm})$ and eluted with water containing increasing amounts of methanol to yield lithospermic acid B (0.5 g) and lithospermic acid (0.3 g). Lithospermic acid B and its salts were identified on the basis of chemical and spectroscopic data reported previously (Tanaka et al 1989). Lithospermic acid, rosmarinic acid and caffeic acid were identified by comparison of their physical and spectroscopic data (Tanaka et al 1989). The chemical structures of these constituents are illustrated in Figure 1.

Measurement of ONOO⁻ scavenging activity

The method of Kooy et al (1994) was used for the determination of ONOO⁻ scavenging activity. Briefly, a stock solution of 5 mM DHR 123 in N. N-dimethylformamide purged with nitrogen was prepared and stored at -20 °C. Each sample was added to a 96-well plate (Nunc. Denmark). A buffer solution of 50 mM sodium phosphate (pH 7.4) containing 90 mM sodium chloride, 5 mM potassium chloride and $100 \,\mu M$ diethylenetriaminepentaacetic acid, and $10 \,\mu\text{M}$ DHR 123 were then added to the plate with or without $10 \,\mu\text{M}$ ONOO⁻. Changes in fluorescent intensity were measured after 5 min. The fluorescence intensity of oxidized DHR 123 was measured with a microplate fluorescence reader (FL 500: Bio-Tex Instruments. USA) at excitation and emission wavelengths of 500 nm and 536 nm, respectively, at room temperature. The scavenging efficacy was expressed as IC50, denoting the concentration of each sample required to induce 50% inhibition of DHR 123 oxidation.

Interaction of lithospermate B with ONOO⁻

To indicate the neutralization mechanism, the interaction of lithospermate B with ONOO⁻ was measured by spectrophotometric analysis as described by Pannala et al (1998). A total of $500 \,\mu\text{M}$ ONOO⁻ in 0.3 M NaOH was added to a solution containing $100 \,\mu\text{M}$ lithospermate B in 50 mM phosphate buffer (pH 7.4), to give a final volume of 1 mL. Each mixed solution was incubated at 37 °C with shaking for 1 h and scanned between 190 nm and 600 nm on an Ultraspec 2000 UV/vis spectrophotometer (Pharmacia-Biotech, UK). The spectral changes in the visible region of lithospermate B in the presence of ONOO⁻ were monitored at 430 nm to determine the existence of nitration. A tube containing no ONOO⁻ was also included for comparative purposes.

Determination of 3-nitrotyrosine

The ability of lithospermate B to suppress ONOO⁻ mediated tyrosine nitration was determined as described above. Briefly, a total of $500 \,\mu\text{M}$ ONOO⁻ in 0.3 M NaOH was added to a solution containing $100 \,\mu\text{M}$ tyrosine in the presence of $100 \,\mu\text{M}$ lithospermate B in 50 mM phosphate buffer (pH 7.4), to give a final volume of 1 mL. The formation of 3-nitrotyrosine by the reaction of tyrosine with ONOO⁻ was also included for comparative purposes. The spectrum of the peak displayed at 430 nm indicates the formation of 3-nitrotyrosine.

Sample preparation for the detection of nitrated proteins

A 2.5- μ L sample of lithospermate B dissolved in 10% EtOH was added to 95 μ L BSA (0.5 mg protein mL⁻¹) or LDL (1 mg protein mL⁻¹). The mixed samples were incubated with shaking at 25 °C for 1 h. Then, 2.5 μ L of ONOO⁻ (final concn: 100 μ M for BSA, 200 μ M for LDL) in 0.3 M NaOH was added. The samples were further incubated with shaking for 30 min at 25 °C.

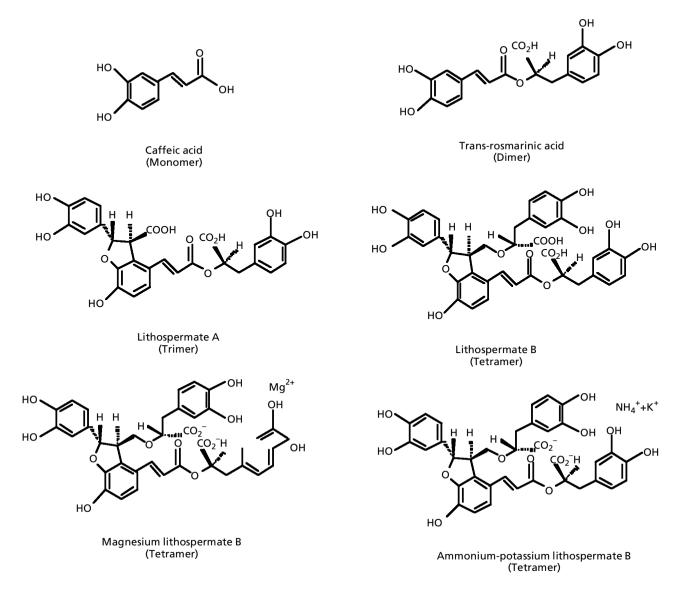


Figure 1 Structures of active components from Salvia miltiorrhiza.

Determination of nitrated proteins

The prepared samples in the gel loading buffer, pH 6.8, (0.125 M Tris(hydroxymethyl)aminomethane, 4% sodium dodecyl sulfate, 20% glycerol, 10% 2-mercaptoethanol and 0.2% bromophenol blue) in a ratio of 1:1 were boiled for 5 min. A total of 20 μ L of each sample was separated on a sodium dodecyl sulfate polyacrylamide mini-gel (10% for BSA, 6% for LDL) at 100 V and transferred to a polyvinylidene fluoride membrane at 100 V for 1.5 h in a wet transfer system (Bio-Rad, Hercules, CA, USA). The membrane was immediately placed in a blocking solution (10% non-fat dry milk in TBS-T buffer containing 10 mM Tris, 100 mM NaCl and 0.1% Tween 20, pH 7.5) at 4°C overnight. The membrane was washed in TBS-T buffer for 30 min and then incubated with a monoclonal anti-nitrotyrosine antibody (diluted 1:2000 in TBS-T buffer) at

room temperature for 2 h. After two 15-min washings in TBS-T buffer, the membrane was reacted with a horseradish peroxidase-conjugated anti-mouse secondary antibody from sheep (0.1% non-fat milk, diluted 1:2000 in TBS-T buffer) at room temperature for 2 h. After three 15-min washings in TBS-T buffer, antibody labelling was detected using enhanced chemiluminescence and exposed to radiographic film. Prestained blue protein markers were used for molecular weight determination.

Statistical analysis

Data are expressed as the mean \pm s.d. Results were analysed by a one-factor analysis of variance. Differences between the means of individual groups were assessed by Fischer's protected LSD post-hoc test.

Iractions from Salvia multiormiza.	
Extract	Activity (%)
Crude extract	65.7 ± 1.9
Fraction A	45.5 ± 0.4
Subfraction A-1	8.7 ± 0.9

744 + 01

 73.5 ± 0.3

 29.2 ± 0.8

 Table 1
 ONOO⁻ scavenging activity of crude extract and its fractions from *Salvia milliorrhiza*.

Values are mean \pm s.d.

Subfraction A-2

Subfraction A-3

Fraction B

Results

ONOO⁻ scavenging effects of *S. miltiorrhiza* crude extract and its fractions

The ONOO⁻ scavenging effects of *S. miltiorrhiza* and its fractions were measured using the DHR 123 method. The crude *S. miltiorrhiza* extract showed 65.7% scavenging activity, and its water-soluble fraction A and lipid-soluble fraction B showed scavenging activities of 45.5% and 29.2%, respectively. Fraction A was further subfractionated into A-1, A-2 and A-3. As seen in Table 1, the subfraction A-2, eluted with 50% aqueous MeOH, showed the highest ONOO⁻ scavenging activity (74.4%).

ONOO⁻ scavenging activity of six components isolated from subfraction A-2

Six active components purified using a water/EtOH mixture from subfraction A-2 were identified (see Figure 1). Table 2 presents the IC50 values of active ONOO⁻ scavenging components measured at various concentrations (0.2, 0.4, 1.6, 3.2 and 6.4 μ M). IC50 values were 1.34 μ M for caffeic acid, 0.92 μ M for trans-rosmarinic acid, 1.11 μ M for lithospermate A, 0.60 μ M for magnesium lithospermate B, 0.60 μ M for ammonium-potassium lithospermate B, and 0.55 μ M for lithospermate B. Among these components, lithospermate B, a tetramer of caffeic acid, showed the most potent scavenging efficacy.

Interaction of lithospermate B with ONOO⁻

Scanning of the reaction mixture produced spectra of lithospermate B with and without ONOO⁻ (data not shown). No peak was observed at 430 nm, indicating no nitration (data not shown). The absence of nitration suggested that the interaction of ONOO⁻ with lithospermate B was mediated by an electron donation reaction.

Effect of lithospermate B on ONOO⁻ mediated 3-nitrotyrosine

The spectrum of the reaction of tyrosine with ONOO⁻ was observed. Absorbance at 430 nm indicated the form-

Table 2 ONOO⁻ scavenging activity of six components isolated from subfraction A-2.

Active component	IC ₅₀ (µM)
Caffeic acid	1.34 ± 0.11
Trans-rosmarinic acid	0.92 ± 0.12
Lithospermate A	1.11 ± 0.18
Lithospermate B	0.55 ± 0.03
Magnesium lithospermate B	0.60 ± 0.03
Ammonium-potassium lithospermate B	0.60 ± 0.02

ation of 3-nitrotyrosine, which disappeared when lithospermate B was incubated with tyrosine before the addition of ONOO⁻, implying that lithospermate B blocked the formation of 3-nitrotyrosine (data not shown).

Effect of lithospermate B on nitration of BSA and LDL by ONOO⁻

To investigate the action of lithospermate B on BSA and LDL nitration by ONOO⁻, nitrotyrosine levels were measured using Western blotting (Figure 2). In the absence of ONOO⁻, BSA and LDL showed no evidence of nitrotyrosine

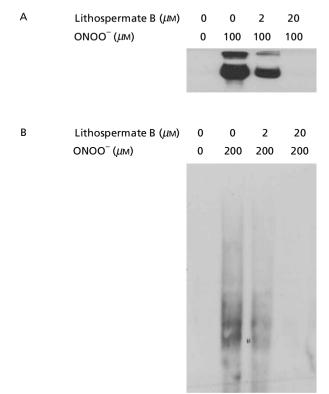


Figure 2 Effect of lithospermate B on nitration of bovine serum albumin (A) and low-density lipoprotein (B) by ONOO⁻.

formation. Although the addition of $ONOO^-$ to BSA or LDL induced nitrotyrosine formation, the pre-incubation of lithospermate B with BSA or LDL led to reduced nitrotyrosine levels in a dose-dependent manner. The protection of lithospermate B against $ONOO^-$ was found to be effective.

Discussion

The inhibition of O_2^- and/or NO generation or the direct scavenging of O_2^- and/or NO are viable approaches to reducing ONOO⁻ induced damage (Arteel et al 1999). So far, the prevention strategies against ONOO⁻ damage are organized into two general approaches, one of which is to prevent the formation of ONOO⁻. Previous studies on S. miltiorrhiza showed that when used as a herbal treatment, patients with non-insulin dependent diabetes (Jiang et al 1997) and coronary heart disease (Xing et al 1996) experienced increased superoxide dismutase levels and decreased lipid peroxidation. Glutathione peroxidase and catalase activities were also enhanced in chronic cor pulmonale patients pretreated with S. miltiorrhiza (Zhang & Chen 1994). In addition, S. miltiorrhiza was shown to suppress NO production and to scavenge $O_2^$ during xanthine and xanthine oxidase reactions in animals with ischaemia/reperfusion injury (Kuang et al 1996; Zhao et al 1996). Additional information on S. miltiorrhiza comes from studies of isolated components. For example, in mice, lithospermate B protected liver macrophages against D-galactosamine and lipopolysaccharide-induced tumour necrosis factor and O_2^- production (Hase et al 1997).

Magnesium lithospermate B, a salt form of lithospermate B, is reported to have strong antioxidant effects (Yokozawa et al 1997). Rats given magnesium lithospermate B after a subtotal nephrectomy showed restoration of superoxide dismutase and catalase activities to almost normal levels and a concomitant decrease in the hydroxyl radical level (Yokozawa et al 1997). Thus, *S. miltiorrhiza* and its component lithospermate B seem to be effective inhibitors and scavengers of both nitrogen- and oxygenderived reactive species.

The second preventative strategy against ONOO⁻ is to inactivate ONOO⁻ itself. The present study showed that ONOO⁻ was effectively scavenged by *S. miltiorrhiza* and its fractions. The six active components extracted from *S. miltiorrhiza* (caffeic acid, trans-rosmarinic acid, lithospermate A, lithospermate B, magnesium lithospermate B and ammonium-potassium lithospermate B) were shown to scavenge ONOO⁻. In the present study, lithospermate B was found to be the most effective ONOO⁻ scavenger among the six components. Other studies have shown that caffeic acid, the basic moiety of lithospermate B, has significant ONOO⁻ scavenging ability (Pannala et al 1998).

Hydroxyl groups are known to be major contributors in scavenging ONOO⁻. Most of the antioxidant activities of flavonoids, hydroxycinnamic acids and catechin polyphenols seem to depend on hydroxyl groups for ONOO⁻ scavenging activity (Haenen et al 1997; Chung et al 1998; Pannala et al 1998). Therefore, hydroxyl groups and double bonds might play an important role in scavenging ONOO⁻.

An important finding of the present study was the data showing the possible reaction mechanism of lithospermate B when exposed to ONOO⁻. ONOO⁻ scavengers are shown to react with ONOO⁻ via two possible pathways: nitration or electron donation (Pannala et al 1998). Evidence obtained from the UV scan showed the reaction of lithospermate B with ONOO⁻ produced no absorbance at 430 nm, implying that lithospermate B acts as an electron donor to neutralize ONOO⁻. On the other hand, nitration, as observed in the conversion of tyrosine into 3-nitrotyrosine by ONOO⁻, showed peak absorbance at 430 nm. Lithospermate B likely decreased the availability of ONOO⁻, which resulted in blocking the formation of 3-nitrotyrosine by ONOO⁻. This possibility was confirmed by evidence of no increased absorbance at 430 nm, when lithospermate B was exposed to ONOO⁻ in the presence of tyrosine in the same concentration as lithospermate B.

The sources of ONOO⁻ are wide-ranging. For example, stimulated macrophages, neutrophils, lymphocytes and endothelial cells are known to generate ONOO⁻ (Deliconstantinos et al 1995: Hukkanen et al 1997: Gagnon et al 1998). Therefore, both the membrane and cytosol are potential targets for ONOO⁻ attack. This is true also for proteins; their constituent amino acids, in particular, are among the preferred target molecules. Nitration reactions and modifications by ONOO⁻ predominantly result from the nitration of tyrosine residues in proteins (Smith et al 1997; He et al 1999; Luoma & Yla-Herttuala 1999). In our study, BSA, a standard protein, was converted to nitrotyrosine-BSA by ONOO-, and lithospermate B inhibited the albumin nitration. The formation of nitrotyrosine-LDL by ONOO⁻ was also effectively blocked by lithospermate B.

Nitrated LDL is implicated in atherosclerosis (Panasenko et al 1997). In-vitro, nitrated LDL is known to be taken up and degraded by macrophages, which can lead to massive cholesterol deposition and foam cell formation, essential steps in lesion development (Panasenko et al 1997). It is suggested that macrophage cells have a macrophage scavenger receptor that is able to recognize chemically nitrated as well as oxidized LDL (Graham et al 1993). In this regard, the finding from the present study that pretreatment with lithospermate B reduces the formation of nitrated BSA and LDL is important. Also noteworthy is the finding that lithospermate B blocks the formation of nitration at a concentration as low as $20 \,\mu$ M.

In conclusion, it is suggested that lithospermate B exerts an anti-nitration effect by scavenging ONOO⁻. Our results suggest that lithospermate B is a good candidate for a ONOO⁻ scavenging agent to protect the cellular component, tyrosine, against ONOO⁻ via ONOO⁻ scavenging.

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